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Note: Not all of the reactions will occur. For those that do not, write no reaction. 1. A piece of copper is dropped into a container of water. No reaction 2. Liquid bromine is added to a container of sodium iodide crystals. $\text{Br (l)} + 2\text{NaI (s)} \rightarrow 2\text{NaBr (s)} + \text{I}_2\text{(s)}$ 3. An aluminum strip is immersed in a solution of silver nitrate.

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The Ultimate Chemical Equations Handbook SnCO tin(II) carbonate sodium hydrogen carbonate NaHCO 1. 2. 3. 4. 5. 6. 8. 9. 10. 11. vanadium(V) oxide H₂O dihydrogen monoxide (NH) CO ammonium oxalate 4 224 polonium(VI) thiocyanate Po(SCN)₆ tetraphosphorus decaoxide P O 4 10 12. 13. 15. 16. 17. 18. 19. manganese(VII) oxide copper(II) dihydrogen phosphate

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Written by an author with over 38 years of experience in the chemical and petrochemical process industry, this handbook will present an analysis of the process steps used to produce industrial hydrocarbons from various raw materials. It is the first book to

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