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Chapter 1 Section 2 MATTER and Its
Properties Matter □ Anything that has
MASS and VOLUME. ATOMS and
MOLECULES the basic building blocks
of matter. ATOM-- the smallest unit of an
element that still maintains the properties

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that element. ELEMENT is a pure substance made up of only one type of atom.

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Section 1

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question. 1. Atomic radii cannot be
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Properties of Matter pages 70□75 Problem-Solving Lab 1. Explain why the flow of a compressed gas must be controlled for practical and safe use.

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eight 3. nine 4. accepted 5. Dmitri
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atomic number 9. Henry Moseley 10.
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25 Electrons in Atoms Reviewing
Vocabulary Match the definition in
Column A with the term in Column B.
Column A Column B 1. The set of
frequencies of the electromagnetic waves
emitted by the atoms of an element 2. The
minimum amount of energy that can be

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lost or gained by ...

Chapter Assessment

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energy changes (that is, energy produced

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or absorber A system that can exchange
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extensive property. intensive property.
extensive properties (examples) intensive
property (examples) a property that

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Depends on the amount of matter in a sample. a property that depends on the type of matter in a sample not. mass, volume, length, weight.

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Chemistry: Matter and Change □ Chapter
11 31 number of moles of known
substance C H C H C H O O O CO CO
CO H O H O H O number of moles of
unknown substance O CO H O C H CO H
O C H O H O C H O CO use mole ratio 5
mol O 2 1 mol C 3 H 8 3 mol CO 2 1 mol
C 3 H 8 4 mol H 2 O 1 mol C 3 H 8 1 mol

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C 3 H 8 5 mol O 2

MATH SKILLS TRANSPARENCY MASTER 16

A **chemical change** is a change that produces matter with a different composition than the original matter.

Chemical Change A change in which one

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Or more substances are converted into different substances. Heat and light are often evidence of a chemical change. Properties of Compounds Quite different properties than their component elements.

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