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© UCLES 2011 5070/21/M/J/11 For Examiner's Use A4 Hydrogen peroxide, H2O2, is a covalent compound. Hydrogen peroxide decomposes to form water and oxygen. 2H2O2(l) 2H2O(l) + O2(g) (a) Draw a 'dot-and-cross' diagram for a molecule of hydrogen peroxide. [2] (b) The decomposition of hydrogen peroxide involves a change from the liquid state to

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Page 4 Mark Scheme: Teachers' version Syllabus Paper GCE O LEVEL - May/June 2011 5070 21. © University of Cambridge International Examinations 2011. A4 (a) All covalent bond pairs shown (1) Rest of structure correct (1) Ignore inner shell electrons of oxygen [2] (b) Must be a comparison in both marking points Particles in a gas are moving faster than particles in a liquid (1) Particles in a gas are further apart than those in a liquid (1) [2]

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