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12.2: Properties of Liquids and Solids. All liquids evaporate. If volume is limited, evaporation eventually reaches a dynamic equilibrium, and a constant vapor pressure is maintained. All liquids experience surface tension, an imbalance of forces at the surface of

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the liquid.

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CHAPTER 12 REVIEW Liquids and Solids SECTION 12-1

SHORT ANSWER Answer the following questions in the space provided. 1. List the following attractive forces in order of increasing strength: (a) hydrogen bonding (c) ionic bonding (b) London dispersion forces (d) dipole-dipole forces 2. All of the following statements about liquids and gases are true except.

~~12 Liquids and Solids~~

364 CHAPTER 12 FIGURE 12-1 Solids and liquids of different

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densities are shown. The densest materials are at the bottom. The least dense are at the top. (Dyes have been added to the liquids to make the layers more visible.) FIGURE 12-2 Like gases, the two liquids in this beaker diffuse over time. The green liquid food coloring from the drop will eventually form

~~CHAPTER 12 Liquids and Solids~~

Liquids have higher densities than gases and are nearly incompressible. But because their particles are in motion, liquids will diffuse through other liquids. Gases will diffuse faster than liquids due to their particles moving more quickly. Solids will also diffuse, but millions of times more slowly than in liquids. Solids
p.367

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~~Unit 1: Review and Chapter 12: Liquids and Solids Liquids ...~~

Review From Chapter 7 (Point III) At a given temperature, the magnitude of intermolecular forces determines whether a substance is a solid, a liquid, or a gas. ... chemistry chapter 12 liquids and solids 60 Terms. MistahhShail. AP Chemistry Chang Chapter 11 54 Terms. Paula_Widholm. Ch 12 - Liquids and Solids 33 Terms. rdmx.

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Chapter 12 SECTION 1 SHORT ANSWER 1. c 2. a 3. b 2. a.

alcohol b. water c. the gels 3. The mixture is a colloid. The properties are consistent with those reported in Table 3 on page 404 of the text. The particle size is small, but not too small, and the mixture exhibits the Tyndall effect. 4. a.

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Chapter 12: Liquids, Solids and Intermolecular Forces Homework:
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between liquid and solid particles are

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CHAPTER 12. FLUID MECHANICS 170 Because the pressure is the same at all point on the same height $p_0 = F_1 A_1 = F_2 A_2$ (12.20) or $F_2 = A_2 A_1 F_1$. (12.21) More generally one formulate what is known as Pascal ' s law: Pressure

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This video explains the concepts from your packet on Chapter 11 (Liquids and Intermolecular Forces), which can be found here: <https://goo.gl/UhCv2b> Section 1...

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